NOVEL APPROACH TO THE STEREOSELECTIVE SYNTHESIS OF POLYPRENOIS VIA DIRECTED ALDOL CONDENSATION. PREPARATION OF HEPTAPRENOIS WITTCCCOH AND WITTCCTOH.

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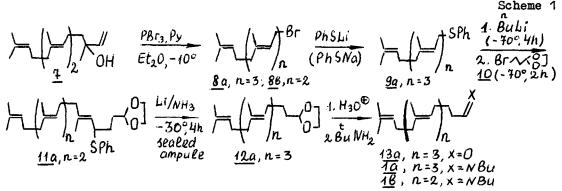
<u>Abstract</u>. New convenient and stereoselective route to linear isoprenoids including the construction of Z-trisubstituted double bond is elaborated based on directed aldol condensation.

We have previously reported¹ that the directed aldol condensation² of λ lithated aldimines <u>1</u> with aldehydes <u>2</u> is highly stereoselective, even when R and R' are longchain alkyles, due to the thermodynamic preference of E-disubstituted acroleins <u>3</u>^{3,4}

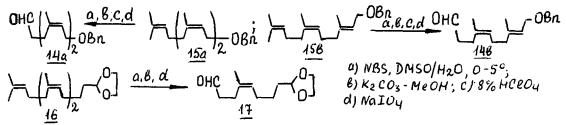
$$R \xrightarrow{NBu} \frac{1. LDA}{2. R'CHO(2)} \xrightarrow{CHO H} \frac{3 \text{ steps}}{R} \xrightarrow{CH_3 H} R'$$

Keeping in mind well documented⁵ reductive transformation of aldehydes $\underline{3}$ into hydrocarbons $\underline{4}$ we supposed that suitably chosen components of the sequence $\underline{1} + \underline{2} - \underline{3} - \underline{4}$ will permit to elaborate new, convenient and stereoselective route to linear isoprenoids containing Z-trisubstituted double bonds. Here we reported the adaptation of this approach for the synthesis of polyprenols $\underline{5}^6$, the membrane soluble carbohydrate carriers essential for the biosynthesis of bacterial polysaccharides as well as of pro- and eucariotic glycoproteins. This communication deals with the synthesis of heptaprenol ω tttcccOH $\underline{5a}$, its Δ^2 -isomer ($\underline{5b}$) and protected aldehyde <u>6</u> which can be used as a substrate for the synthesis of higher polyprenols according to this approach.

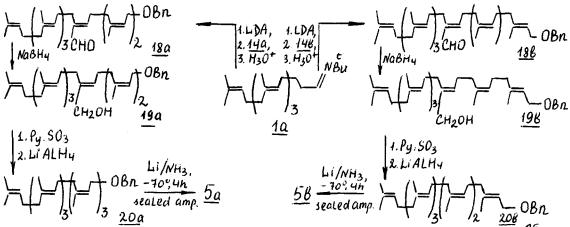
Scheme 1 outlines the sequence of reactions leading to aldimine <u>1a</u> starting from geranyllinalool <u>7</u>. The transformation of <u>7</u> into primary bromide 8a^{7a} proved to be not stereoselective contrary to the earlier claims ^{7,8}. We obtained <u>8a</u> (80%) as the 3:1 mixture of E:Z- Δ^2 -isomers⁹ and this ratio could not be changed by the variations of temperature, solvent and reagent addition order. The reaction of <u>8a</u> with PhSLi or PhSNa produced (87%) the 3:1 mixture of E:Z- Δ^2 -isomers of sulfide 9a¹¹ which was enriched by carefull flash-column chromatography up to the ratio 9:1.¹³ The treatment of \prec -lithated <u>9a</u> with 10 yielded 87% of acetal <u>11a</u>.⁴⁶⁴⁷ Its further desulfurisation (without double bond shift, PMR), hydrolysis of resulting acetal <u>12a</u>^{16,17} and amination of the aldehyde <u>13a</u>¹⁶ lead to aldimine <u>1a</u>¹⁶ with overall yield (from <u>7</u>) about 50%. Aldimine <u>1b</u> was prepared similarly from E,E-farnesylbromide <u>8b</u>.



In order to obtain aldehydes <u>14a,b</u>, the aldehyde components for synthesis of <u>5a,b</u>, the respective benzyl farnesols <u>15a,b</u> were oxidised with good yield using the modified van Tamelen's procedure.¹⁸ Similarly, acetal <u>16</u> (obtained from Z,Z-farnesylbromide in the same manner as <u>12a</u> from <u>8a</u>) was transformed into aldehyde <u>17</u>, the aldehyde component for synthesis of <u>6</u> (Scheme 2). Scheme 2



The crucial steps of the <u>5a,b</u> synthesis were performed according to scheme 3. Thus, the treatment of aldimine <u>1a</u> freshly prepared from 11mM <u>13a</u> with 11mM of LDA in ether-hexan (3:1) solution (1h at -20°, then 1h at 0°, Ar) followed by one-pot reaction of \measuredangle -Li-<u>1a</u> with 8mM of etheral <u>14a</u> (2,5h at-70°, then 2h at -30°) gave after hydrolysis of the reaction mixture (5% aqueous oxalic acid), usual treatment and purification by the flash-column chromatography on silica gel (elution from hexan to ether up to 5 vol % of the latter) the key intermediate $\underline{18a}^{16}$ in about 55% yield and stereochemical purity > 95% (¹H and ¹³C NMR data, cf³). Aldehyde $\underline{18b}^{16}$ was obtained (52%) in the similar manner starting from <u>1a</u> and <u>14b</u>



Aldehydes <u>18a,b</u> were reduced with NaBH₄ into respective alcohols <u>19a,b</u>¹⁶ with about 90% yield. Their hydrogenolysis according to^{5a} furnished benzylic ethers <u>20a,b</u>¹⁶ smoothly debenzilated into desired <u>5a</u> and <u>5b</u>. The total yield of the latters was 12-14% starting from geranyllinalool <u>7</u>. Analogously, condensation of \measuredangle -Li-aldimine <u>1b</u> with <u>17</u> followed by above transformation of CHO-group to methyl gave <u>6</u>, useful synthon for the construction of higher polyprenols according to our scheme.

Thus, the discussed approach might be regarded as an effective route to regular and modified polyprenols for their further biochemical investigations.

References and notes

- 1. N.Ya.Grigorieva, A.V.Semenovsky, Izv.Akad.Nauk, Ser.Khim. 1976, 2644.
- 2. G.Wittig, H.D.Frommeld, Chem.Ber. 97, 3548 (1964).
- N.Ya.Grigorieva, E.P.Prokofiev, A.V.Semenovsky, Dokl.AN SSSR, <u>245</u>, 366 (1979).
- 4. K.Ya.Burstein, N.Ya.Grigorieva, Izv.Akad.Nauk, Ser.Khim. 1982, 449.
- 5. a) E.J.Corey, K.Achiwa, J.Org.Chem. <u>34</u>, 3667 (1969); b) J.C.Depezay, Y.Merrer, Tetrahedron Lett., <u>1975</u>, 3469.
- F.W.Hemming, in "MTP International Review of Science. Biochemistry.Ser.1., vol.4. Biochemistry of Lipids" (T.W.Goodwin Ed.), Butterworths, London -University Park Press, Baltimore, 1974, p.39; V.N.Shibaev, Uspekhi biol. Khimii, <u>17</u>, 187 (1976).
- 7. a) L.Ruzicka, G.Firmenich, Helv.Chim.Acta, <u>22</u>, 392 (1939); b) O.Isler, R. Ruegg, L.Chopard-dit-Jean, H.Wagner, K.Bernhard, ibid, <u>39</u>, 897 (1956).

- 8. M.Kadama, Y.Matsuki, S.Ito, Tetrahedron Lett., 1975, 3065.
- 9. The Δ^2 -isomer ratio of <u>8a</u> was estimated by the integration of proton signals (PMR-250 MHz) at δ 1,72 ppm (cis-CH₃-C³) and δ 1,78 ppm (trans-CH₃-C³), cf¹⁰, and at δ 4,01 ppm and δ 3,95 ppm (CH₂Br in E- and Z- Δ^2 -<u>8a</u>, respectively).
- 10. Y.Naruta, J.Org.Chem. 45, 4097 (1980).
- 11. The Δ^2 -isomer ratio of <u>9a</u> was estimated by integration of proton signals at δ^1 1,59 ppm (cis-CH₃-C³) and 1,75 ppm (trans-CH₃-C³)¹² and at 3,56 ppm and 3,58 ppm (CH₂S in E- and Z- Δ^2 -<u>9a</u>, respectively).
- 12. J.F.Biellmann, J.B.Ducep, Tetrahedron, <u>27</u>, 5861 (1971).
- 13. Fortunately the complete resolution of \triangle^2 -E:Z mixture of <u>9a</u> is not necessary. The \triangle^2 -Z-contamination in <u>9a</u> resulted only in ω ttccccOH-contamination in the desired heptaprenol ω tttcccOH. This admixture should not affect the biochemical properties of the latter compound, since, according to V.N.Shibaev and others¹⁴, the moraprenol (<u>5</u>, m=3, n=7) is effectively incorporated in the biosynthesis of polyprenols by Salmonella enzymes whose own prenols have structure <u>5</u> (m=2, n=8).
- 14. V.N.Shibaev, Yu.Yu.Kusov, T.N.Druzhinina, N.A.Kalinchuk, N.K.Kochetkov,
 V.A.Kilesso, S.Sh.Rozhnova, Bioorgan.Khim., <u>4</u>, 47 (1978).
- 15. G.Büchi, H.Wüest, J.Org.Chem., <u>34</u>, 1122 (1969).
- 16. All new compounds gave satisfactory elemental analysis and spectral (IR, MS, ¹H- and ¹³C-NMR) data.
- 17. The Δ^5 -isomer mixture of acetals <u>11a</u> and <u>12a</u> were analysed by ¹³C NMR. According to these data the alkylation of <u>9a</u> and the subsequent steps are completely stereoselective.
- 18. E.E. van Tamelen, T.J.Curphey, Tetrahedron Lett., 1962, 121.

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